Synthesis of Desethanoqinghaosu, a Novel Analogue of the Antimalarial Qinghaosu

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Desethanoqinghaosu, a novel potential antimalarial qinghaosu analogue containing the unique C-O-O-C-O-C=O (or cyclic peroxide lactone) moiety present in the natural product, has been synthesized from R-(+)-citronellal and its structure and stereochemistry have been verified from spectral data and a single-crystal X-ray analysis.

Two total syntheses^{1,2} of qinghaosu [(1), arteannuin or artemisinin], a clinically useful antimalarial sesquiterpene lactone endoperoxide isolated from the Chinese drug Qinghao (*Artemisia annua*),^{3,4} have recently been reported. Our previous studies⁵ on structure–activity relationships and a programme of synthesis directed at the preparation of more active, simple analogues of (1) have led to the hypothesis that a requirement for the cyclic peroxide function to display antimalarial activity might involve the unique C–O–O–C–O–C–O–C–O–C–O–C–O–C–O–C–O–C=O moiety found in (1). We now report a stereoselective synthesis of desethanoqinghaosu (2), the first example of a simple analogue containing this structural feature.

As shown in Scheme 1, the synthesis of (2) involved an initial conversion of (R)-(+)-citronellal (3) into a dihydroxy



compound (4) by known procedures.^{6.7} Oxidation (70%) of (4) with Jones reagent followed by methylation (85%) with CH₂N₂ gave a ketomethyl ester (5). Wittig reaction of (5) with MeOCH₂PPh₃Cl and BuⁿLi in dry Et₂O afforded an enol methyl ether (6) (40%). Photo-oxygenation⁸ of a tetrahydrofuran (THF) solution of (6) with MeCHO in the presence of Rose Bengal at -70 to -78 °C under a bubbling stream of oxygen yielded a mixture of diastereoisomers [(7a) and (7b), (35%)], which were separated by h.p.l.c. (MeCN: H₂O = 8:2, Nucleosil 100-C₁₈, 7 µm). The structures of (7a) and (7b) were elucidated by ¹H n.m.r. analysis,[†] and the stereochemistry of the former was confirmed by a single-crystal

[†] Spectroscopic data. (7a): m.p. 88–89 °C (from CHCl₃/n-hexane); ¹H n.m.r.: δ 0.89 (3H, d, J 6.5 Hz, Me-5), 1.26 (3H, d, J 5.4 Hz, Me-7), 1.27 (3H, d, J 7.7 Hz, Me-9), 3.40 (3H, s, OMe-8), 3.62 (3H, s, CO₂Me-9), 3.80 (1H, dq, J 7.7 and 2.9 Hz, H-9), 4.79 (1H, s, H-8), and 5.61 (1H, q, J 5.5 Hz, H-7).

⁽⁷b): oil; ¹H n.m.r.: δ 0.88 (3H, d, J 6.4 Hz, Me-5), 1.21 (3H, d, J 7.1 Hz, Me-9), 1.36 (3H, d, J 5.7 Hz, Me-7), 3.48 (3H, s, OMe-8), 3.67 (3H, s, CO₂Me-9), 5.04 (1H, s, H-8), and 5.40 (1H, q, J 5.7 Hz, H-7).

^{(2):} m.p. 95–96 °C (from MeOH/n-hexane); ¹H n.m.r.: δ 1.01 (3H, d, J 6.3 Hz, Me-5), 1.18 (3H, d, J 7.3 Hz, Me-9), 1.34 (3H, d, J 5.4 Hz, Me-7), 2.75 (1H, ddd, J 13.5, 3.9, and 1.8 Hz, H-2), 3.04 (1H, dq, J 7.3 and 3.9 Hz, H-9), 5.63 (1H, s, H-8), and 5.66 (1H, q, J 5.4 Hz, H-7).



Scheme 1. Reagents and conditions: i, $ZnBr_2$; ii, B_2H_6 , $NaOH/H_2O_2$; iii, Jones reagent; iv, CH_2N_2 ; v, Bu^nLi , $MeOCH_2PPh_3Cl$; vi, $^{1}O_2$, Rose Bengal, THF, hv, -70 to -78 °C; vii, $^{1}O_2$, Rose Bengal, MeOH, hv, -70 to -78 °C; viii, HCl gas; ix, 60% HClO₄.

X-ray analysis (Figure 1).‡ Photo-oxygenation of a methanolic solution of (6) with MeCHO in the presence of oxygen and Rose Bengal at -70 to -78 °C followed by HCl gas treatment

‡ Crystal data. (7a): C₁₅H₂₆O₆, M = 302.37, monoclinic, space group $P2_1$, a = 15.589(6), b = 9.494(4), c = 11.748(4) Å, $\beta = 109.76(3)^\circ$, U = 1636.3 Å³, Z = 4, $D_c = 1.227$ g cm⁻³, μ (Cu- K_{α}) = 7.4 cm⁻¹. (2): C₁₃H₂₀O₅, M = 256.30, monoclinic, space group C2, a = 19.944(6), b = 6.419(1), c = 10.721(2) Å, $\beta = 93.31(2)^\circ$, U = 1370.2 Å³, Z = 4, $D_c = 1.242$ g cm⁻³, μ (Cu- K_{α}) = 7.5 cm⁻¹.

Intensity data [3104 and 1317 unique forms for (7a) and (2), respectively] were recorded on an Enraf-Nonius CAD-4 diffractometer (Cu- K_{α} radiation, $\lambda = 1.5418$ Å; incident-beam graphite monochromator; $\omega - 2\theta$ scans, $\theta_{max} = 67^{\circ}$). Data were corrected for the usual Lorentz and polarization effects; a decay correction was also necessary for (7a) which was unstable to X-irradiation. Both crystal structures were solved by direct methods. Initial non-hydrogen atom positions were derived from E-maps. Hydrogen atoms, save those on the ester methyl group of (7a), were located in difference Fourier syntheses. Full-matrix least-squares refinement of atomic positional and thermal parameters (anisotropic C, O; calculated H positions) converged at R = 0.094, $R_w = 0.114$ for (7a) and R = 0.036, $R_w =$ 0.049 for (2) over 2022 and 741 reflections, respectively, with I > $3.0 \sigma(I)$. Atomic co-ordinates, thermal parameters, and bond lengths and angles, have been deposited with the Cambridge Crystallographic Data Centre. See Notice to Authors, Issue No. 1.



Figure 1. Structure and solid-state conformation of one of the molecules of (7a) in the asymmetric crystal unit; calculated hydrogen atom positions [C(15) hydrogens not defined by the analysis] are represented by small circles.



Figure 2. Structure and solid-state conformation of (2); small circles denote calculated hydrogen atom positions.

and acidic hydrolysis with 60% $HClO_4$ produced the target compound [(2), (15%)]. The complete structure and stereochemistry of (2) were elucidated by single-crystal X-ray analysis,‡ and ¹H n.m.r. spectroscopy.† Bioassay of (2), (7a), and (7b) as potential antimalarial agents is in progress.

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